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Explanation of Anomalous Gas-Chromatographic (Gc) Behaviour in a Homologous Series of 2-Chloroethyl Phosphonic Acid Dialkyl Esters

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EXPLANATION OF ANOMALOUS GAS-CHROMATOGRAPHIC (GC) BEHAVIOUR IN A HOMOLOGOUS SERIES OF 2-CHLOROETHYL PHOSPHONIC ACID DIALKYL ESTERS

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The study of GC separation of 2-chloroethylphosphonic acid di-n-alkyl (1-5 C atoms) esters (synthesised by us) on silicone stationary phases (OV-1, OV-17, OV-225) revealed a deviation from the expected linear dependence of retention indexes (RI) versus the

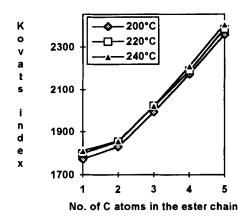


FIGURE 1 Kovats indexes versus no. of C atoms in the ester chain for OV-225 column

	TABLE I Conformations and						lues
	R	R Most stable conformations				μ(D)	
		R	R	C-Ethyl	Cl	25°C	200°C
7	CH3	sc(±sp)	±sp	±sp(±ap) a	ac(±ap)	2.01	2.20
C	2H5	±sp(sc)	±sp	±sp(±ap)	ac(±ap)	2.06	2.08
<u>C</u>	3H7	±sp(sc)	±sp	±sp(±ap)	ac(±ap)	2.08	2.09

number of C atoms of the alkyl chain: the first member of the series presents stronger retention than one can expect. This anomalous behaviour was observed especially on polar stationary phase (OV-225, see Figure 1), and was emphasised of the column increase temperature. In an attempt to rationalise the above mentioned facts, we tried to relate the RI values to a global polarity parameter: the dipole moment, μ . The μ for $ClC_2H_4P(O)(OR)_2$ calculated by a method described in [1] (tested by comparing the calculated µ values with experimental ones for alkyl phosphonic acid dialkylesters), using molecular mechanics (COSMIC package) in the search of the conformational space, AM1 method (MOPAC 6.0) for the μ values of the conformers, and Boltzmann distribution for the global value (see Table I). At low temperature, the μ values are not related to the Kovats indexes. Those calculated at 200°C (column temperature range) demonstrate that, indeed, only in

the case of the methyl derivative, the temperature rising led to a higher μ (enhanced population of the more polar conformers: ac position for the C-Ethyl group - ac for the Cl, $\mu \approx 2.7$ D, or $\pm ap$ for one R, $\mu \approx 3.8 \div 4.3$ D). It can be concluded that dipole-dipole forces contribute to the separation process of the first members of the series.

[1] L. Kurunczi, T. Sulea and T.I. Oprea, J. Mol. Struct. (Theochem), 306, 93 (1994)